

Kinetics of batch microbial degradation of phenols by indigenous *Pseudomonas fluorescence*

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ABSTRACT: The potential of various organisms to metabolize organic compounds has been observed to be a potentially effective means in disposing of hazardous and toxic wastes. Phenols and their compounds have long been recognized as one of the most recalcitrant and persistent organic chemicals in the environment. The bioremediation potential of an indigenous *Pseudomonas fluorescence* was studied in batch culture using synthetic phenol in water in the concentration range of (100–500) mg/L as a model limiting substrate. The effect of initial phenol concentration on the degradation process was investigated. Phenol was completely degraded at different cultivation times for the different initial phenol concentrations. Increasing the initial phenol concentration from 100 mg/L to 500 mg/L increased the lag phase from 0 to 66 h and correspondingly prolonged the degradation process from 84 h to 354 h. There was decrease in biodegradation rate as initial phenol concentration increased. Fitting data into Monod kinetic model showed the inhibition effect of phenol. The kinetic parameters have been estimated up to initial phenol concentration of 500 mg/L. The r_{smax} decreased and K_s increased with higher concentration of phenol. The r_{smax} has been found to be a strong function of initial phenol concentration. The culture followed substrate inhibition kinetics and the specific phenol consumption rates were fitted to Haldane, Yano and Koga, Aiba *et al.*, Teissier and Webb models. Between the five inhibition models, the Haldane model was found to give the best fit. Therefore, the biokinetic constants estimated using these models showed good potential of the *Pseudomonas fluorescence* and the possibility of using it in bioremediation of phenol waste effluents.

Key words: Biodegradation, kinetic model, batch cultivation, bioreactor, concentration

INTRODUCTION

Phenol and its derivatives is the basic structural unit in a wide variety of synthetic organic compounds (Annadurai *et al.*, 2000). It is an organic, aromatic compound that occurs naturally in the environment (Prpich and Daugulis, 2005), but is more commonly produced artificially from industrial activities such as petroleum processing, plastic manufacturing, resin production, pesticide production, steel manufacturing and the production of paints and varnish (Mahadevaswamy *et al.*, 1997; Bandyopadhyay *et al.*, 1998). This aromatic compound is water-soluble and highly mobile (Collins and Daugulis, 1997) and as such waste waters generated from these industrial activities contain high concentrations of phenolic compounds (Chang *et al.*, 1998) which eventually may reach down

to streams, rivers, lakes, and soil, which represent a serious ecological problem due to their widespread use and occurrence throughout the environment (Fava *et al.*, 1995). Phenol is a listed priority pollutant by the U.S. Environmental Protection Agency (EPA, 1979) and is considered to be a toxic compound by the Agency for Toxic substances and Disease Registry (ATSDR, 2003). The adverse effects of phenol on health are well documented (Calabrese and Kenyon, 1991) and death among adults has been reported with ingestion of phenol ranging from 1 to 32 g (Prpich and Daugulis, 2005). The low volatility of phenol and its affinity for water make oral consumption of contaminated water the greatest risk to humans (Prpich and Daugulis, 2005). A variety of techniques involving physical, chemical and biological methods have been used for the removal of phenol from industrial effluents and contaminated

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waters with bioremediation receiving the most attention due to its environmental friendliness, its ability to completely mineralize toxic organic compounds and of low-cost (Kobayashi and Rittman, 1982; Prpich and Daugulis, 2005). Microbial degradation of phenol with different initial concentrations ranging from 50-2000 mg/L have been actively studied using shake flask, fluidized-bed reactor, continuous stirred tank bioreactor, multistage bubble column reactor, air-lift fermenter and two phase partitioning bioreactor methods (Bettmann and Rehm, 1984; Sokol, 1988; Annadurai *et al.*, 2000; Reardon *et al.*, 2000; Ruiz-ordaz *et al.*, 2001; Oboirien *et al.*, 2005; Prpich and Daugulis, 2005; Saravanan *et al.*, 2008) and these studies have shown that phenol can be aerobically degraded by wide variety of fungi and bacteria cultures such as *Candida tropicalis* (Ruiz-ordaz *et al.*, 2001, Chang *et al.*, 1998; Ruiz-ordaz *et al.*, 1998); *Acinetobacter calcoaceticus* (Paller *et al.*, 1995); *Alcaligenes eutrophus* (Hughes *et al.*, 1984; Leonard and Lindley, 1998); *Pseudomonas putida* (Hill and Robinson, 1975; Kotturi *et al.*, 1991; Nikakhtari and Hill, 2006); and *Burkholderia cepacia*

G4 (Folsom *et al.*, 1990, Solomon *et al.*, 1994). In microbial degradation of phenol under aerobic conditions, the degradation is initiated by oxygenation in which the aromatic ring is initially monohydroxylated by a mono oxygenase phenol hydroxylase at a position *ortho* to the pre-existing hydroxyl group to form catechol. This is the main intermediate resulting from metabolism of phenol by different microbial strains. Depending on the type of strain, the catechol then undergoes a ring cleavage which can occur either at the *ortho* position thus initiating the *ortho* pathway that leads to the formation of succinyl Co-A and acetyl Co-A or at the *meta* position thus initiating the *meta*-pathway that leads to the formation of pyruvate and acetaldehyde. Feist and Hegeman (1969), Hill and Robinson (1975) and Leonard and Lindley (1998) have described the biodegradation or metabolism of phenol by *Pseudomonas putida*, *Pseudomonas cepacia*, and *Alcaligenes eutrophus* respectively via the *meta* cleavage pathway; while, Paller *et al.* (1995) described the biodegradation of phenol by *Acinetobacter calcoaceticum* via the *ortho* cleavage pathway.

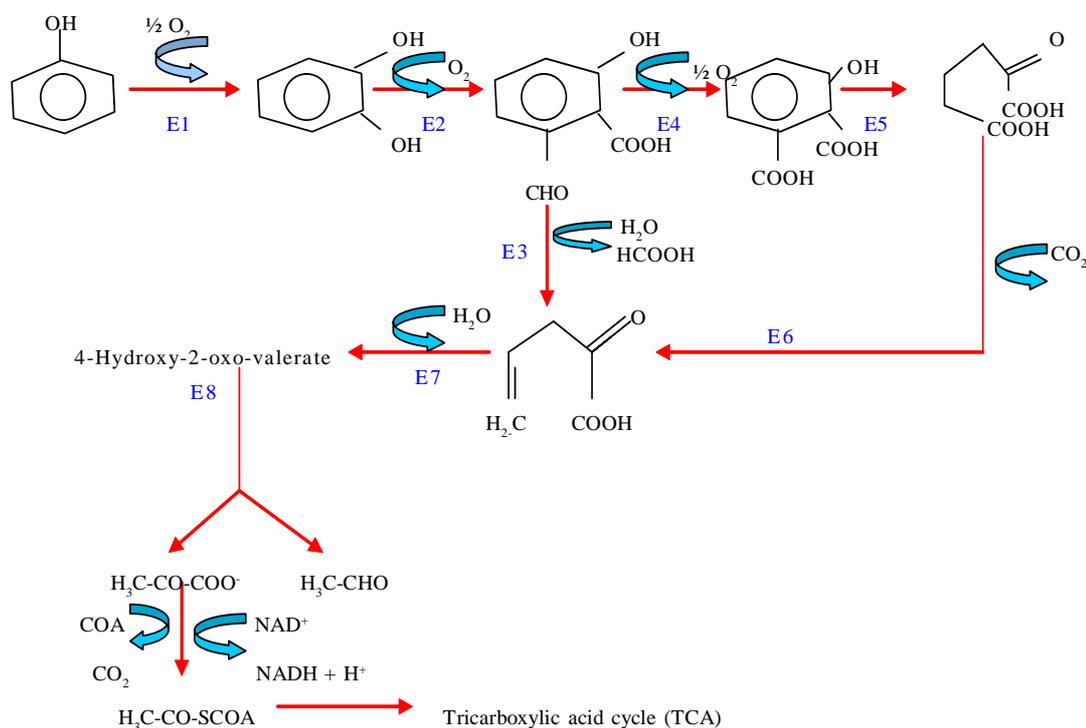


Fig. 1: The meta-cleavage pathway for the biodegradation of phenol

A= Phenol, B= Catechol, C= 2-Hydroxymuconic semialdehyde, D= 2-Hydroxymuconate, E= 2- Oxo- 4- enoate, F= 2- Oxo-penta-4-enoate, G= Pyruvate, H= Acetaldehyde, I= Acetyl Co A, E1= Monooxygenase phenol hydroxylase, E2=Catechol-2, 3- dioxygenase, E3= Hydrolase, E4= Dehydrogenase, E5= Isomerase, E6= Decarboxylase, E7= Hydrotase, E8= Aldolase

The *meta* cleavage pathway for the biodegradation of phenol as presented by Feist and Hegeman (1969) is as shown in Fig. 1. A variety of kinetic substrate utilization and inhibition models have been used to describe the dynamics of microbial growth on phenol. Of these various models, the Monod and Andrew (Haldane) equations has been extensively used to describe phenol biodegradation (Bandyopadhyay *et al.*, 1998; Reardon *et al.*, 2000; Oboirien *et al.*, 2005). The Monod (1949) and Haldane (Andrew, 1968) equations are based on the specific growth rate, but may also be related to the specific substrate consumption rate (Edwards, 1970; Solomon *et al.*, 1994). Other kinetic models have been propagated. Sokol (1988) has reported a better fit for a modified Monod-Haldane equation while Schroeder *et al.* (1997) have shown a better fit for Yano and Koga equation amongst the tested inhibition models. In spite of the rather extensive use of phenol biodegradation processes, surprisingly, little work has been published on phenol microbial degradation kinetics based on specific substrate consumption rate using pure or mixed cultural systems. The objective of this present study was to investigate the degradation potential of an indigenous (local isolates) *Pseudomonas fluorescence* isolated from an oil-polluted area in Niger-Delta region of Nigeria and the effect of initial phenol (substrate) concentration on its degradation ability.

MATERIALS AND METHODS

The microorganism *Pseudomonas fluorescence* being an indigenous strain isolated from an oil-polluted area in Niger-Delta region of Nigeria was procured from the Department of Microbiology, Obafemi Awolowo University, Ile-Ife, Nigeria. The microorganism was maintained on nutrient agar slant and stored at 4 °C ± 1°C for further use. The mineral salt medium used was modified from the one suggested by Bettman and Rehm (1984). The medium had the following composition per litre: 700 mL deionized water, 100 mL buffer solution A, 100 mL solution B, 50 mL solution C and 50 mL solution D. Compositions of each solution were as follows: Buffer solution A composition K₂ HPO₄ 1.0 g, KH₂PO₄ 0.5 g, (NH₄)₂SO₄ 0.5 g, deionized water 100 mL. Trace element solution B composition NaCl 0.5 g, CaCl₂ 0.02 g, MnSO₄ 0.02 g, CuSO₄·5H₂O 0.02 g, H₃BO₃ 0.01 g, deionized water 50 mL. Solution C composition MgSO₄·7H₂O 0.5 g, deionized water 50 mL, Solution D composition FeSO₄ 0.02 g, Molybdenum powder 0.02

g, deionized water 50 mL. To prevent the precipitation of CaSO₄ and MgSO₄ in storage, the water, buffer solution A; trace elements solution B, solution C and solution D were autoclaved at 121°C for 15 min. After cooling, all the solutions were then mixed together and kept as stock solution from which known quantities were taken for the cultivation of the microorganisms. A primary culture was prepared by transferring two loops full of microorganisms from an agar slant culture into 100 mL of feed medium containing 20 mL of mineral salt medium and 80 mL of 50 mg phenol solution in a 250 mL Erlenmeyer conical flask. This was then incubated in a NewBrunswick gyratory shaker (G25-R model, N.J. U.S.A) for 48 h at a temperature of 30 °C and agitated with a speed of 120 rpm. Thereafter, 10 mL of the primary culture was transferred into another 100 mL of feed medium in a 250 mL Erlenmeyer conical flask and the incubation process was repeated. This was the secondary culture that was used as the inoculums for the degradation studies as this ensures that the organisms had fully adapted to growth on the phenol as sole source of carbon and energy. The experimental studies were carried out in a NewBrunswick Microferm Twin Bioreactor (PH - 22 model, N.J., U.S.A) with 4 litres working volume. 800 mL of the autoclaved mineral salt medium and 3 L of phenol solution (100 mg/L) were measured into the bioreactor vessel and 200 mL of the inoculums was introduced aseptically to make up 4 litres of working volume. The bioreactor was operated for several hours at a temperature of 30 °C, aeration rate of 3.0 vvm and agitation of 300 rpm. Samples were withdrawn at every 6 h for biomass and phenol analysis. This was done in duplicate and average values were used. The undegraded phenol was estimated quantitatively by measuring its absorbance at 510 nm wavelength using UV - Visible Spectrophotometer (Lambda 35, Perkin-Elmer, USA) and 4 - amino antipyrene as color indicator (Yang and Humphrey, 1975). The biomass concentration was estimated using the dry weight method. 50 mL sample of culture broth was withdrawn from the bioreactor and centrifuged (Gallenkamp centrifuge) at 4000 rpm for 20 min in plastic centrifuge tubes. The supernatant was decanted into small bottles and stored at 4 °C for subsequent phenol estimation. The pellets was re-suspended in de-ionized water and re-centrifuged. The supernatant was decanted and pellets rinsed off from the tube into a pre-weighed 1.2 µm pore filter paper (Whatman GF/C). The filter paper saw then dried in

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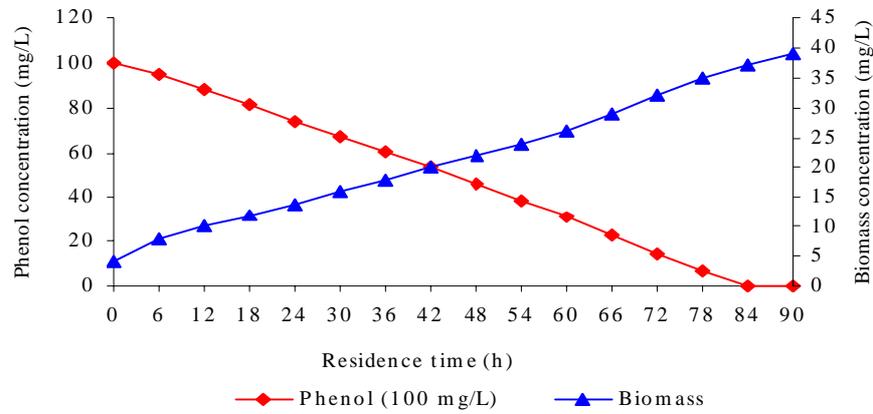


Fig. 2: Experimental data obtained from batch degradation of phenol (100 mg/L) by monoculture of *Pseudomonas fluorescens*

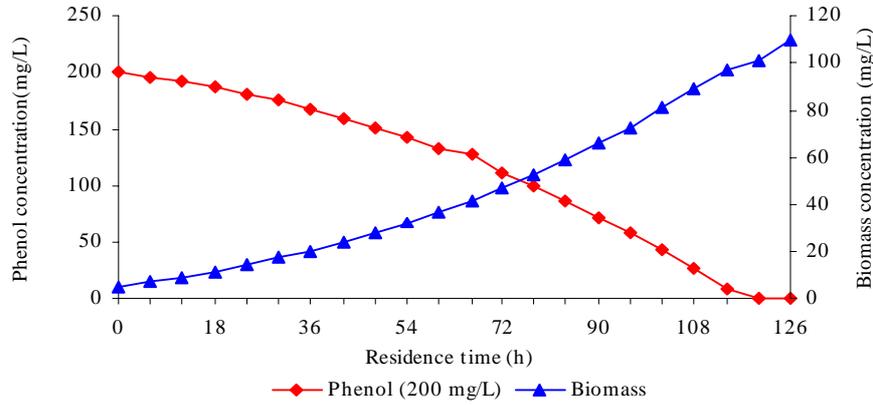


Fig. 3: Experimental data obtained from batch degradation of phenol (200 mg/L) by *Pseudomonas fluorescens*

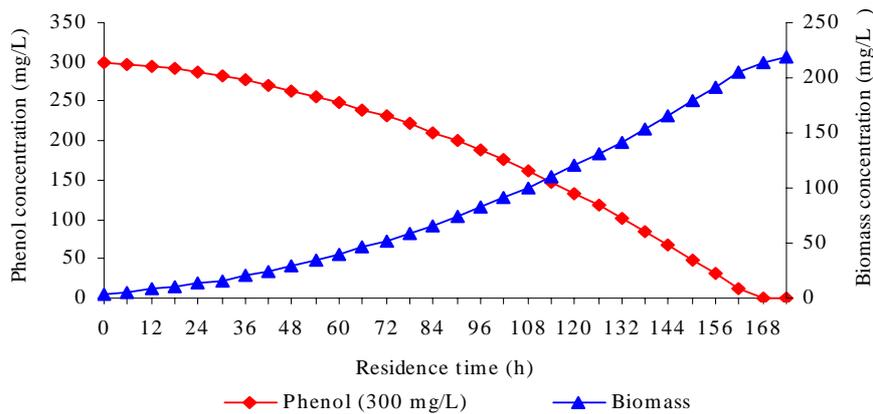


Fig. 4: Experimental data obtained from batch degradation of phenol (300 mg/L) by *Pseudomonas fluorescens*

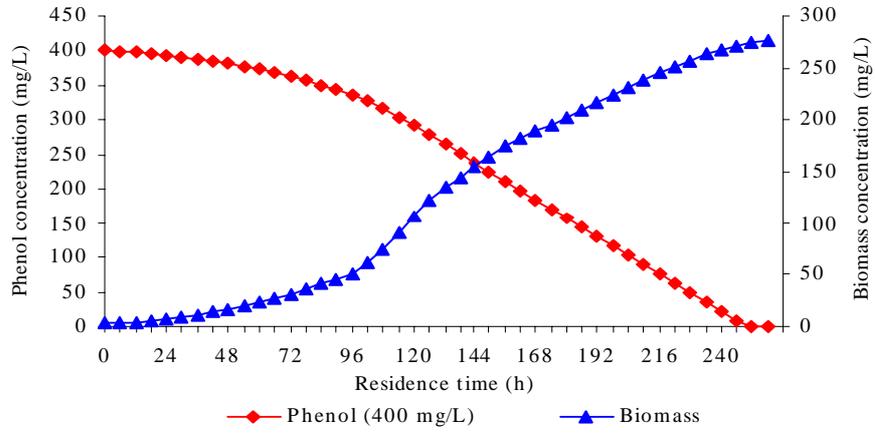


Fig. 5: Experimental data obtained from batch degradation of phenol (400 mg/L) by *Pseudomonas fluorescens*

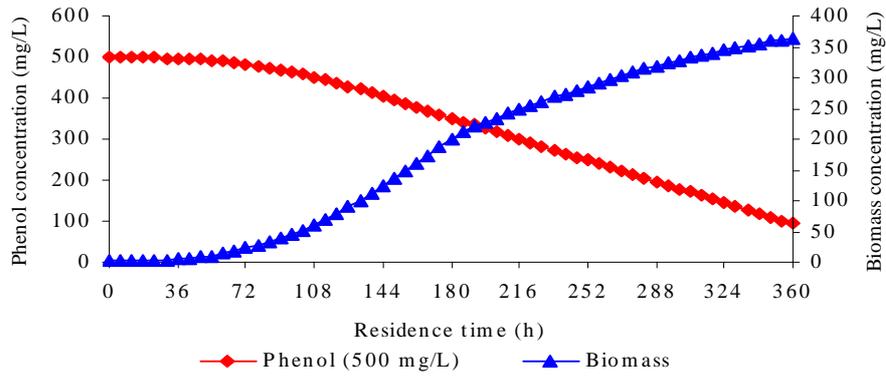


Fig. 6: Experimental data obtained from batch degradation of phenol (500 mg/L) by *Pseudomonas fluorescens*

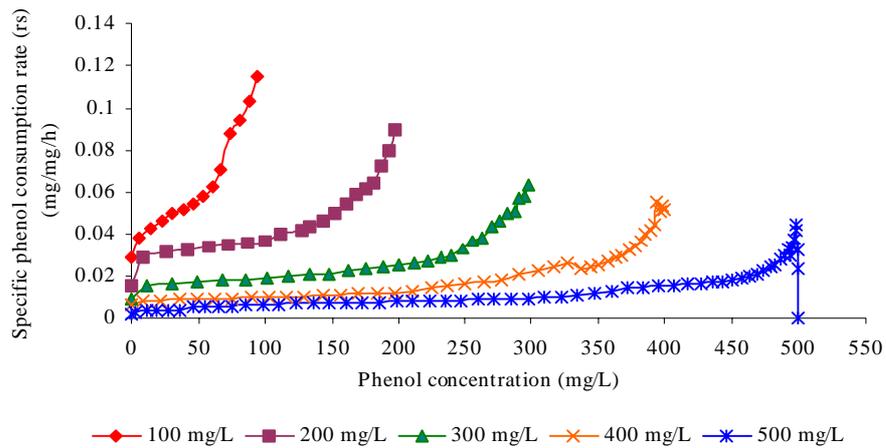


Fig. 7: Specific phenol consumption rate vs phenol concentration for monoculture of *Pseudomonas fluorescens*

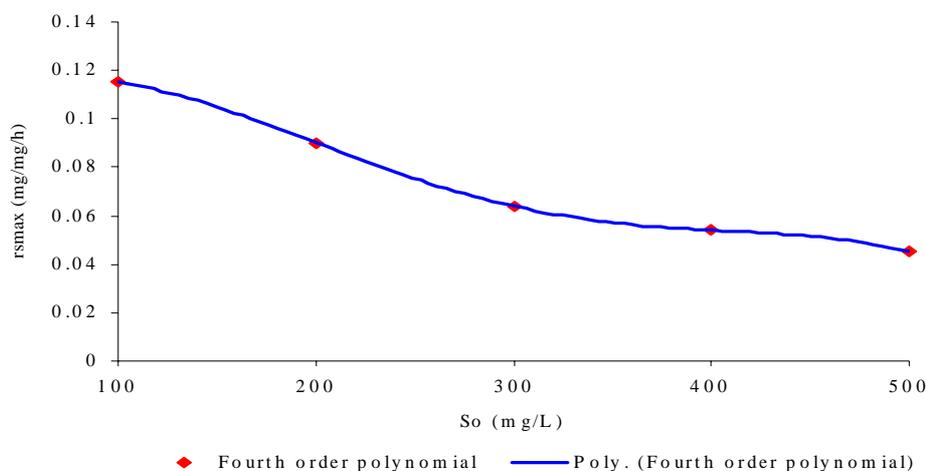


Fig. 8: Variation of r_{smax} with S_o for phenol degradation by monoculture of *Pseudomonas fluorescence*

an oven at 105 °C for between 12-24 h, cooled in a desiccator at room temperature and re-weighed until a constant dry weight was obtained. The difference between the pre-weighed filter paper and the second weight was used to estimate the dry weight of the biomass.

RESULTS AND DISCUSSION

Five batch cultivation experiments were carried out using phenol as single limiting substrate for *Pseudomonas fluorescence*. Different initial phenol concentrations of 100 mg/L to 500 mg/L were used. The extent of phenol degradation using these different initial phenol concentrations was investigated for several batch residence times by intermittent sampling.

Figs. 2 to 6 shows the biodegradation potential of the indigenous *P.fluorescence* in degrading synthetic phenol waste in the concentration range of 100 mg/L to 500 mg/L. Since the degradation proceeds with biomass (cell mass) growth, the Figure also depicts the typical cell growth curve. The cell growth curve has typical exponential and stationary phases with increasing lag phase. It could be seen from Figs. 2 to 6 that the different initial phenol concentrations ranging from (100 – 500) mg/L were completely degraded (consumed) at different residence time of 84 h, 120 h, 168 h, 252 h, 354 h, and during these times the biomass correspondingly increased to a maximum of 37 mg/L, 101 mg/L, 213 mg/L, 274 mg/L, and 362 mg/L respectively. No lag phase was observed for initial phenol concentration of 100 mg/L as shown in Fig. 2. However, for initial phenol concentration of (200, 300,

400, and 500) mg/L corresponding lag phase of 6 h, 18 h, 24 h and 66 h was observed respectively as shown in Figs. 3 to 6. Moreover, it was observed that after the complete degradation or conversion of the different initial phenol concentrations, there was further increase of the biomass at each initial concentration of phenol. This increase may be due to the fact that the added phenol was converted to unknown intermediates that were later consumed. A similar observation has been reported for the isolates of *Pseudomonas* and *Acinetobacter* grown on phenol (Hao *et al.*, 2002; Saez and Rittman, 1991; Prpich and Daugulis, 2005). As shown in Figs. 2 to 6, which indicate the comparison of the time course for phenol substrate consumption of all the five batches, it is evident that the rate of degradation decreased with increase in the initial phenol concentration. Bandyopadhyay *et al.* (1998) and Ruizordaz *et al.* (2001) reported a similar observation on *Pseudomonas putida* and *Candida tropicalis* grown on phenol, respectively. Thus, it is observed that as the initial phenol concentration increased the duration of the lag phase increased; and thereby, prolonging the biodegradation time as a result of decrease in the rate of degradation. This observation is supported by the earlier works of Andrews (1968), Hill and Robinson (1975), Collins and Daugulis (1997), and Oboirien *et al.* (2005). According to Prpich and Daugulis (2005), the rate of substrate consumption was suggested to be the most important measure of microbe performance. Zilli *et al.* (1993) gave a similar report. Relatively very few data exists in the literature on this parameter. Most of the data available concerns specific growth rate. It

Table 1: Kinetic constants obtained from the fitting of experimental batch phenol degradation runs data for *P.fluorescence* to Monod kinetic model

Initial phenol concentration S_0 (mg/L)	Half-saturation constant K_s (mg/L)	Maximum specific phenol consumption rate r_{smax} (mg/mg/h)
100	31.1	0.115
200	88.7	0.090
300	176.8	0.064
400	294.8	0.054
500	556.4	0.045

Table 2: Kinetic parameters value obtained from five kinetic models fitted to the biodegradation data of *P.fluorescence*

Model	r_{smax} (mg/mg/h)	K_s (mg/L)	K_1 (mg/L)	K_2 (mg/L)	K_i (mg/L)	R^2	RMSE
Yano & Koga	38.3	105.9	0.384	31.2	-	0.7824	0.0300
Teissier	0.694	0.727	-	-	0.394	-3.35	0.1001
Aiba <i>et al.</i>	0.694	0.727	-	-	0.394	-2.649	0.1001
Webb	6.283	-3.369	-1.493	-	54.6	0.7837	0.0299
Haldane	0.357	50.2	-	-	67.7	1.000	0.0003

was on this basis that the specific phenol (substrate) consumption rate was calculated and plotted against phenol concentration as shown in Fig. 7. As seen from Fig. 7, the specific phenol consumption rate (r_s) decreased as the phenol concentration (S) decreased for each of the different initial phenol concentrations. Therefore, it seems that there is also an influence of the initial phenol concentration on the specific phenol consumption rate. Hinteregger *et al.* (1992) and Abd-El Hameidshalaby (2003) reported a similar observation. According to Layokun *et al.* (1987) the growth of microorganisms corresponds to the degradation (consumption) of the substrate. Hence, the growth of microorganisms on phenol can be described by the most commonly used kinetic models that can be based on specific substrate consumption rate as proposed by Posten (Solomon *et al.*, 1994) and which have been used by Zilli *et al.* (1993) and Schroeder *et al.* (1997). In this work, phenol was completely degraded and kinetic model of Monod (1949) based on specific substrate consumption rate was used to evaluate the degradation of phenol at different initial concentration by the indigenous *P. fluorescence*. The classical method of obtaining kinetic parameters (constants) is to linearize kinetic models. Recently, non-linear least squares computer fitting of data to model equations has been used (Schroeder *et al.*, 1997; Reardon *et al.*,

2000; Saravanan *et al.*, 2008). The non-linear least square fitting routine of MATLAB 6.5 software package was used to fit the Monod kinetic model to the different batch experimental data. The parameters of Monod (K_s and r_{smax}) was fitted to the experimental calculated specific phenol consumption rate and the corresponding phenol concentration under the constraint that r_s never exceeds the maximum obtainable specific consumption rate (r_{smax}) and the results are presented in Table 1. As seen from the Table, the Monod parameters (r_{smax} and K_s) show a definite trend of variation. The r_{smax} decreased and K_s increased as the initial phenol concentration increased. Considering the fact that K_s is inversely related to the affinity of the microbial system for the substrate (Pirt, 1975), this increase of K_s corresponds to a decrease in affinity of the bacteria (*P. fluorescence*) for phenol. It is therefore evident that inhibition becomes prominent as initial phenol concentration increased. Thus, the r_{smax} is a strong function of initial phenol concentration (S_0). The variation of r_{smax} with S_0 has been indicated in Fig.8 and also fitted by the fourth order polynomial fit from which r_{smax} at any value of S_0 within the range of 100-500 mg/L of phenol concentration may be predicted. However, the observation of substrate inhibition due to phenol can be modeled using substrate inhibition models described in literature (Schroeder *et al.*, 1997).

The experimental results of specific phenol consumption rate variation with initial phenol concentration was fitted to five inhibition models of Haldane (Andrew, 1968), Yano and Koga (1969), Aiba *et al.* (1968), Teissier (Edwards, 1970), and Webb (Edwards, 1970). The model with the best fit was selected on the basis of highest correlation coefficient (R^2) and the least root mean square error (RMSE). The kinetic parameters of the models (r_{smax} , K_s and K_i) were estimated using the non – linear regression routine of MATLAB 6.5, and the results along with the R^2 and RMSE are given in Table 2. The results revealed that between the five models, Haldane has a correlation coefficient (R^2) greater than 0.90 and the least root mean square error (RMSE = 0.0003), and thus, indicating a very good fit to the batch experimental data. In this work, the biodegradation potential of phenol by *Pseudomonas fluorescense* has been examined. It was shown that the organism completely degraded phenol up to a maximum concentration of 500 mg/L in mineral salt medium, and that, initial phenol concentration has effect on the rate of degradation. The higher the initial phenol concentration, the longer is the lag phase of the culture, and the lower the rate of degradation; thus, the slower the biodegradation becomes. The culture followed substrate inhibition kinetics as indicated by the kinetic parameters value of Monod model equation, in which r_{smax} decreased and K_s increased, as initial phenol concentration increased. Thus, the experimental batch biodegradation data was fitted to Haldane, Yano and Koga, Aiba *et al.*, Teissier, and Webb, substrate inhibition models. It was shown that of all the models, the Haldane model represent the biodegradation data reasonably very well, and hence, may be proposed as the best model to describe the phenol biodegradation behaviour of *P. fluorescense*.

The performance of this indigenous *P. fluorescense* is being compared with well-known effective degraders of phenol with emphasis on maximum specific substrate consumption rate. Reported values of the maximum specific substrate consumption rate (r_{smax}) varied from 0.001 to 2.6/h (Folsom *et al.*, 1990; Zilli *et al.*, 1993; Schroeder *et al.*, 1997). Folsom *et al.* (1990) and Schroeder *et al.* (1997) reported an r_{smax} value of 2.6 hr⁻¹ and 0.4 hr⁻¹ for *Pseudomonas cepacia* G4 respectively. While, Zilli *et al.* (1993) reported a value of 0.0016/h for *Pseudomonas putida* NCIMB 10015. The r_{smax} value of 0.357 mg/mg/h for *P. fluorescense* was relatively comparable to that of *Pseudomonas cepacia* G4 (r_{smax}

= 0.4/h), however greater than that of *P. putida* NCIMB 10015. In conclusion, the biokinetic constants evaluated using the models showed good tolerance and growth of the indigenous organism, and therefore, a complete degradation of the phenol substrate. The performance of the indigenous strain in biodegradation of phenol in the nutrient medium is good.

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Nomenclature

- K_s Half-saturation constant (mg/L)
 K_i Inhibition constant (mg/L)
 r_s Specific phenol (substrate) consumption rate (mg/mg/h.)
 r_{smax} Maximum specific phenol (substrate) consumption rate (mg/mg/h.)
 S, C_s Substrate (phenol) concentration (mg/L)

$$\text{Monod (1949): } r_s = \frac{r_{s,max} S}{K_s + S}$$

$$\text{Haldane (Andrews, 1968): } r_s = \frac{r_{s,max} S}{K_s + S + \frac{S^2}{K_i}}$$

$$\text{Aiba } et al. (1968): r_s = r_{s,max} \frac{C_{s,exp} \left(-\frac{C_s}{K_i} \right)}{K_s + C_s}$$

Teissier (Edwards, 1970):

$$r_s = r_{s,max} \left[\exp \left(-\frac{C_s}{K_i} \right) - \exp \left(-\frac{C_s}{K_s} \right) \right]$$

$$\text{Webb (Edwards, 1970): } r_s = r_{s,max} \frac{C_s \left(1 + \frac{C_s}{K_i} \right)}{K_s + C_s + \frac{C_s^2}{K_1}}$$

Yano and Koga (1969):

$$r_s = r_{s,max} \frac{C_s}{K_s + C_s + \frac{C_s^2}{K_1} + \frac{C_s^3}{K_2}}$$

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